

Mechanisms at Structural Phase Transitions

H. Böhm

Institut für Geowissenschaften
Universität Mainz
Germany

V. Kahlenberg, J. Kusz, O. Pawlig, M. Riester

Definition

Structural Phase Transition:

It occurs in a crystalline solid with a sudden change in symmetry because of a change of pressure and/or temperature.

A change of the symmetry is associated with the change of the structure.

classification

◆ First order phase transitions.

at the temperature T_c : $\Delta S \neq 0$

$$\Delta V \neq 0$$

Mechanisms:

nucleation and growth,

range of coexisting phases

characteristic features:

thermal hysteresis

classification

◆ Second order phase transitions.

at the temperature T_c : $\Delta S = 0$

$$\Delta V = 0$$

Mechanisms:

continuous transition

characteristic feature:

no thermal hysteresis;

Landau Theory is applied

Example

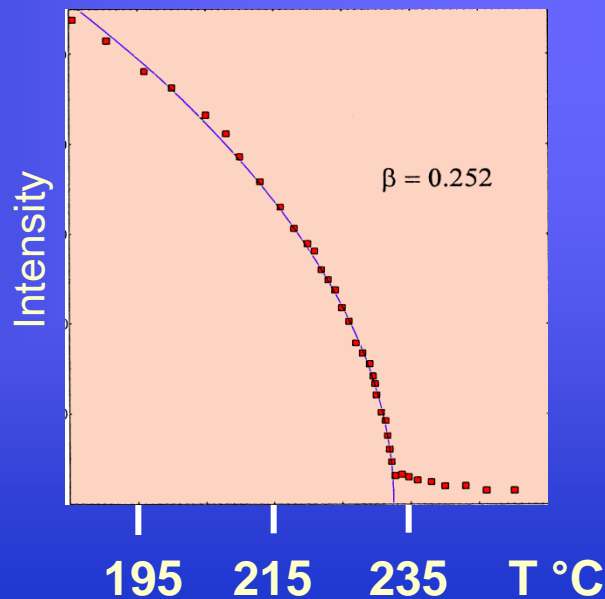


Sequence of phases

α - Phase
C2/c

$$c_0(\alpha) = 2 c_0(\beta)$$
$$T_c = 233^\circ\text{C}$$

β - Phase
C2/m



$$\sqrt{I} \sim \eta = \eta_0 (T_c - T)^\beta$$

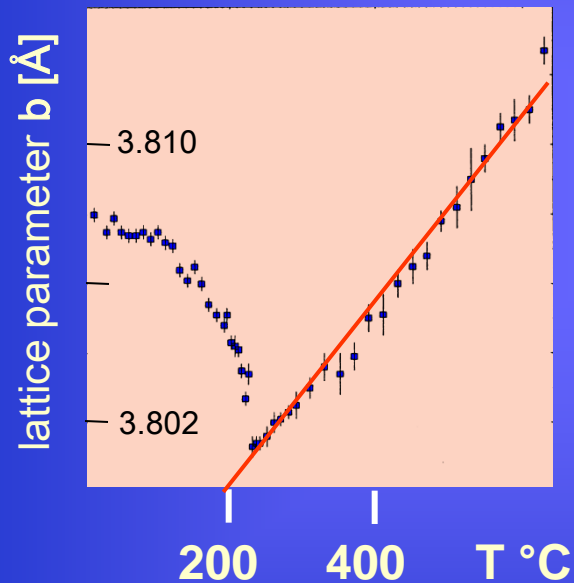
η = order parameter

Superstructure
reflection

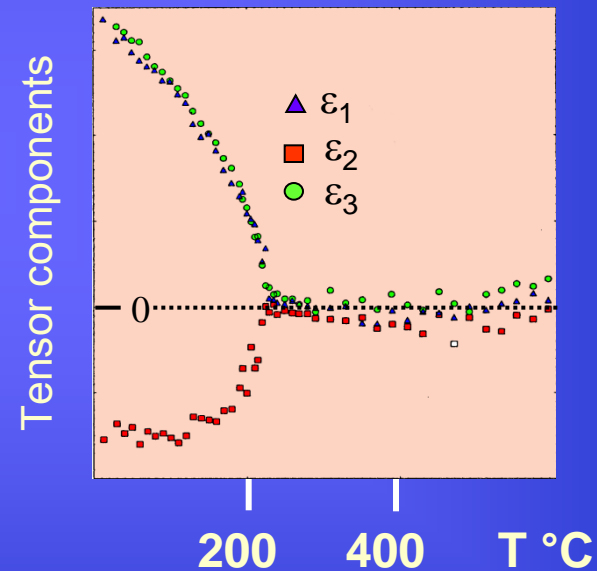
continued



Lattice parameter **b**



Spontaneous strain

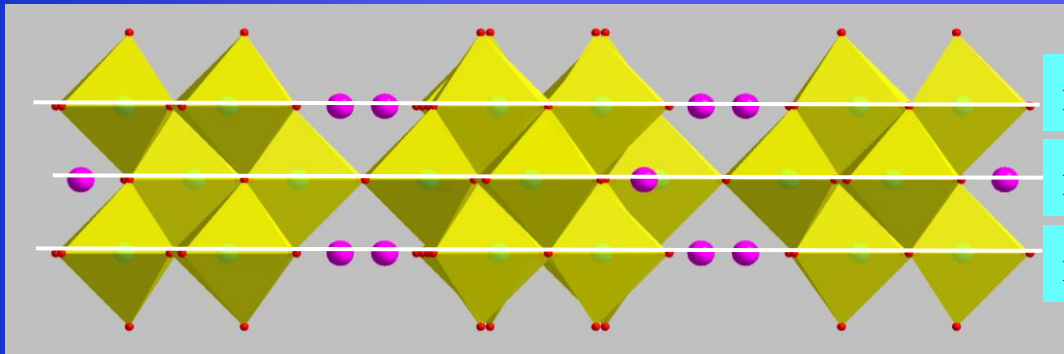


No thermal hysteresis:
2.nd order phase transition ,
coupling between η and the spontaneous
strain

continued



structure : β -Phase



m

m

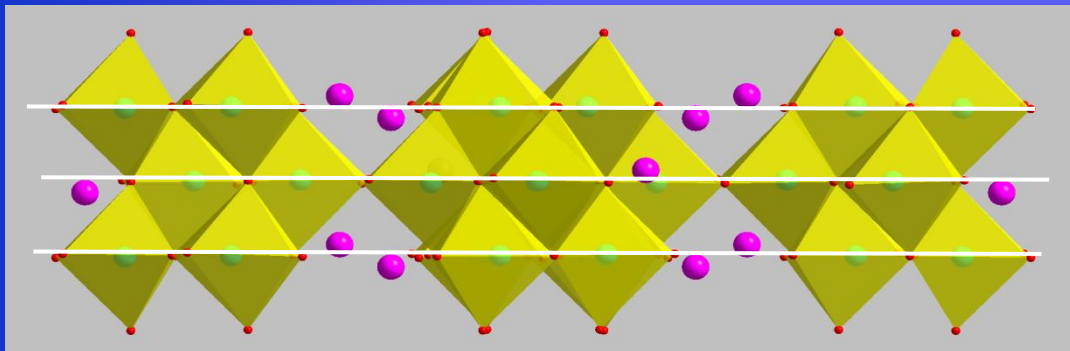
m

Interpretation in terms of the Landau Theory:

At T_c a continuous shift of the Bi-atoms begins off the mirror plane.

structure : α -Phase

Bi - Atome



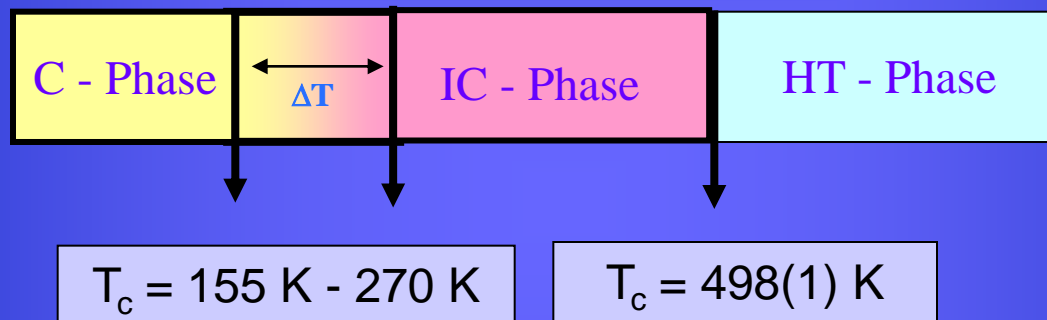
Comment:

The transition is triggered by the ordering of the lone electron pair of Bi.

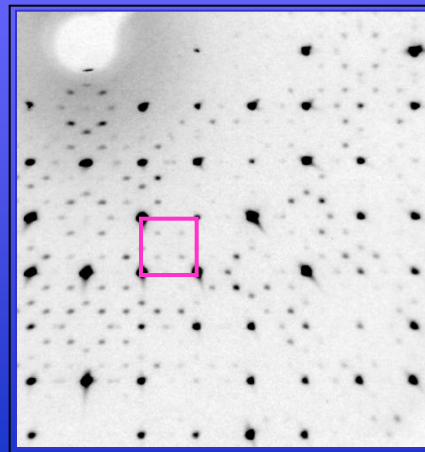
Example



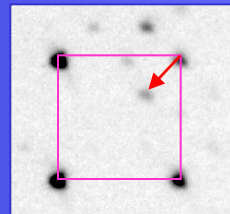
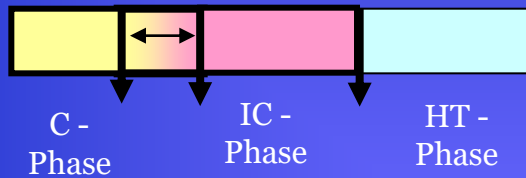
Sequence of phases:



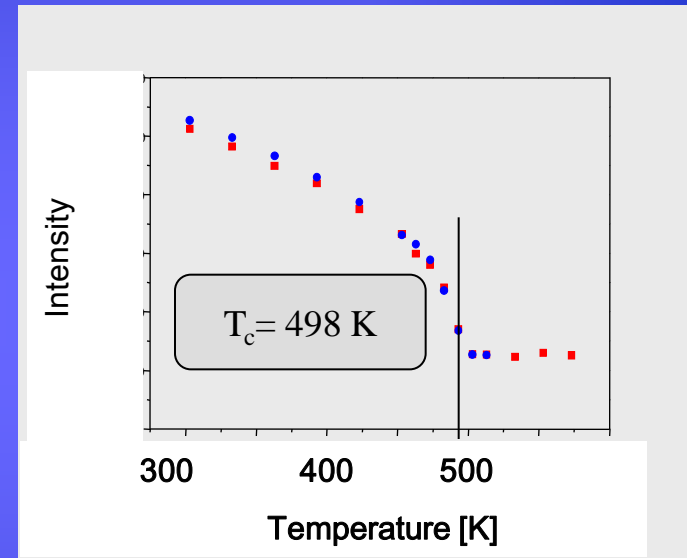
Diffraction pattern
at RT:



continued



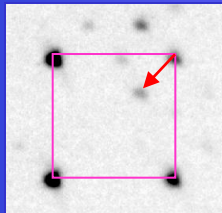
Satellite-Intensity



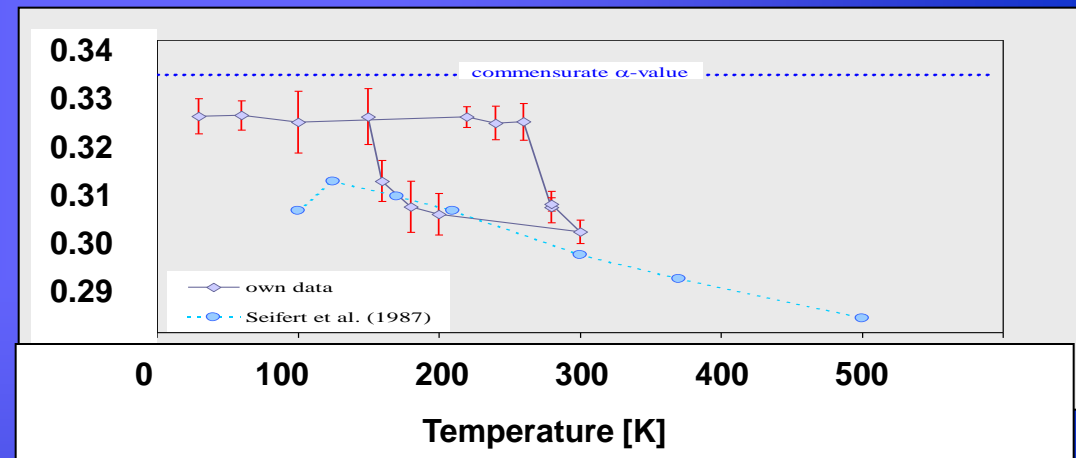
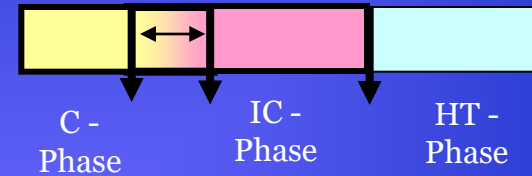
Phase transition at about 500 K

No thermal hysteresis:
2.nd order phase transition

continued



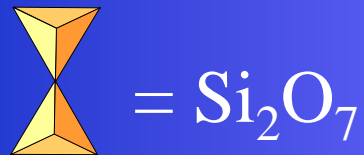
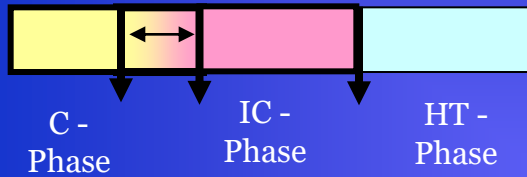
Variation of the q-vector
(incommensurate phase)



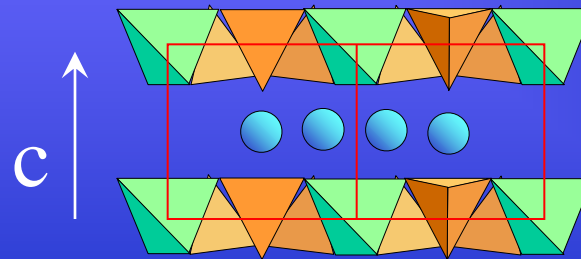
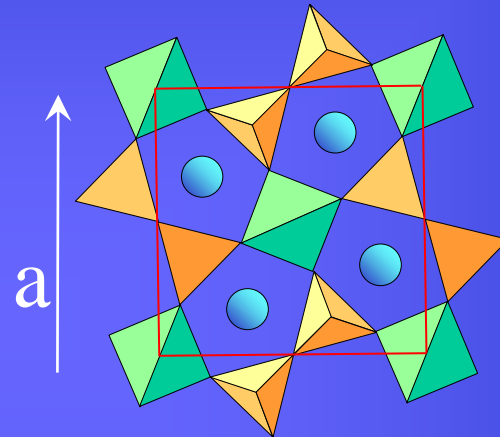
Phase transition
between 155 K und 270 K

Thermal hysteresis:
1. order phase transition

$\text{Ca}_2\text{CoSi}_2\text{O}_7$ continued



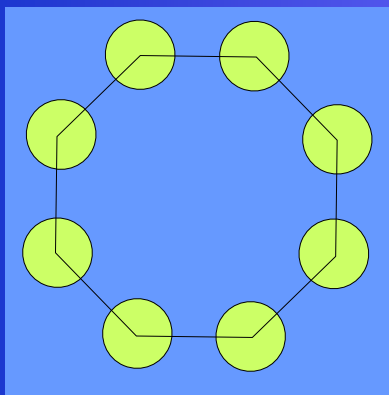
HT - Phase



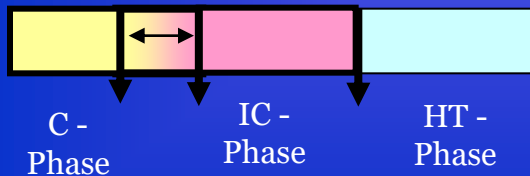
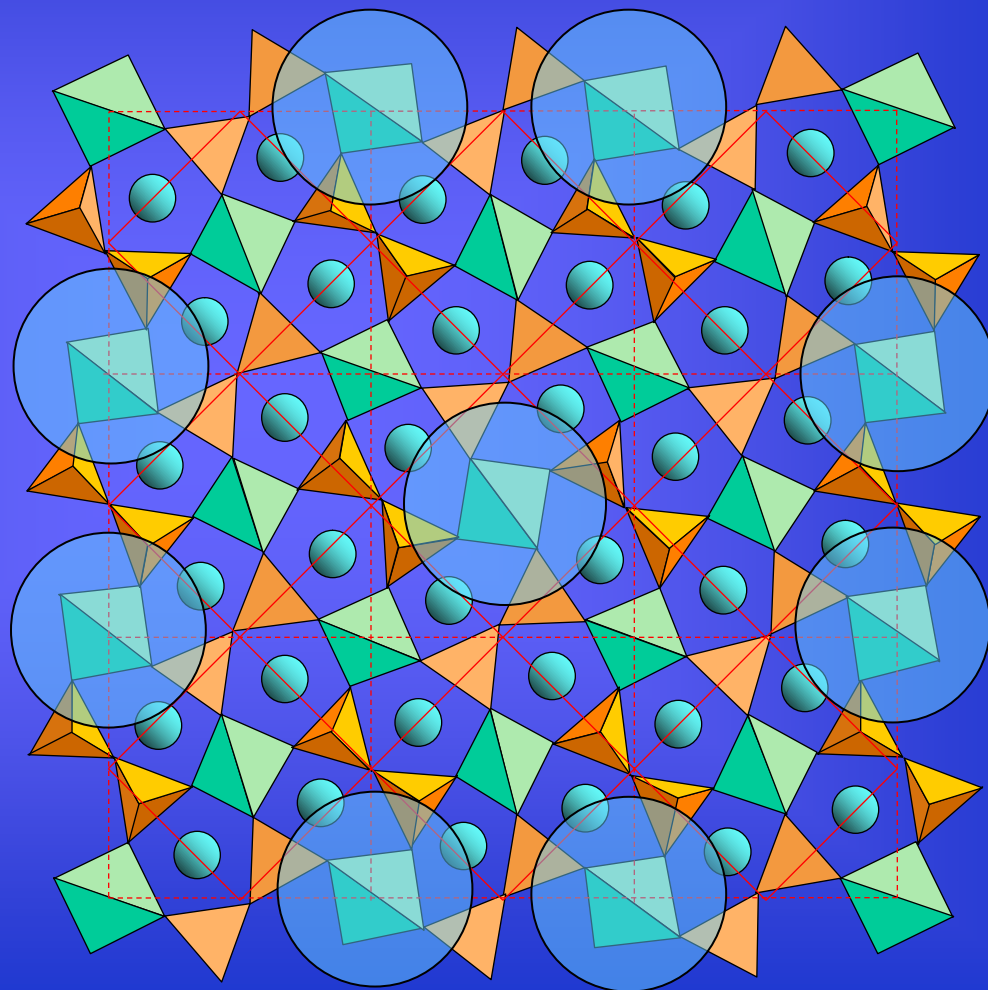
continued



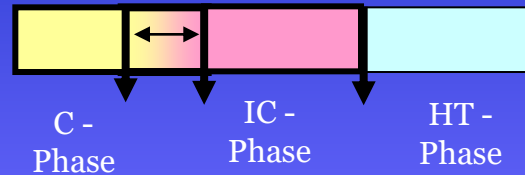
C - Phase



3·a

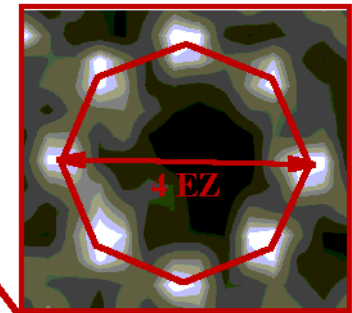
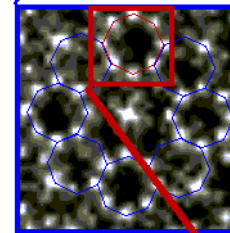
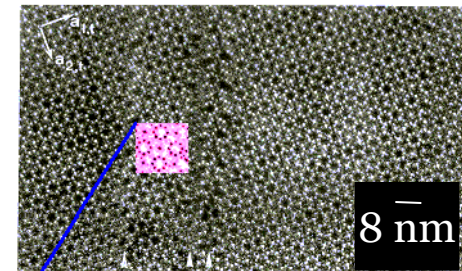
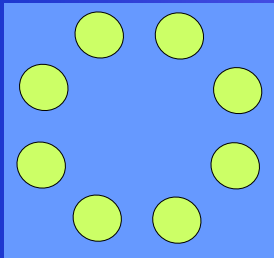


continued



IC - Phase

HRTEM picture at RT:
Octagonal rings.

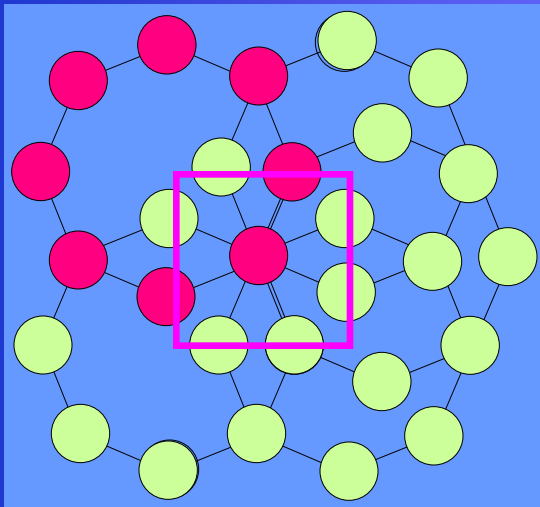
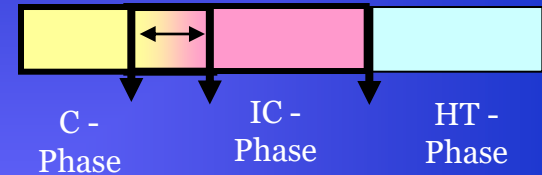


Van Heurk et al., 1992

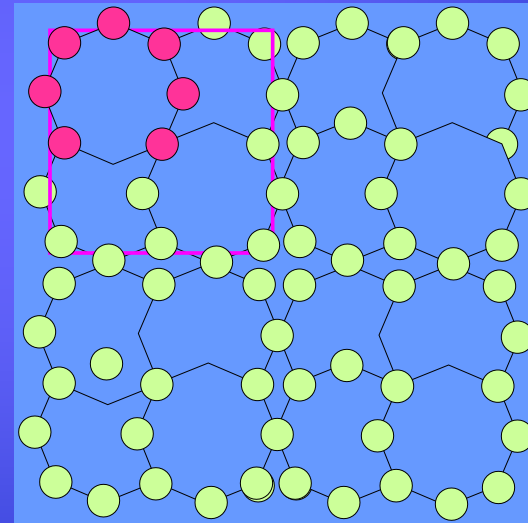
continued



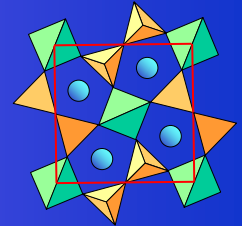
Relationship between the phases



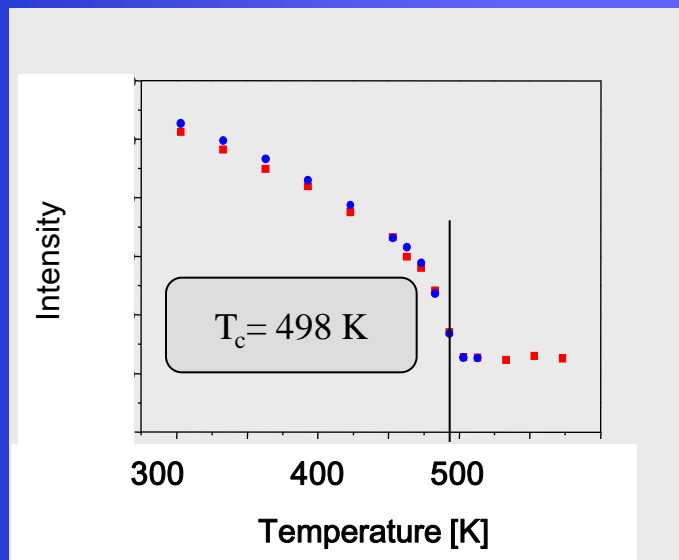
C-Phase:
An ordered superposition
of octagonal rings.



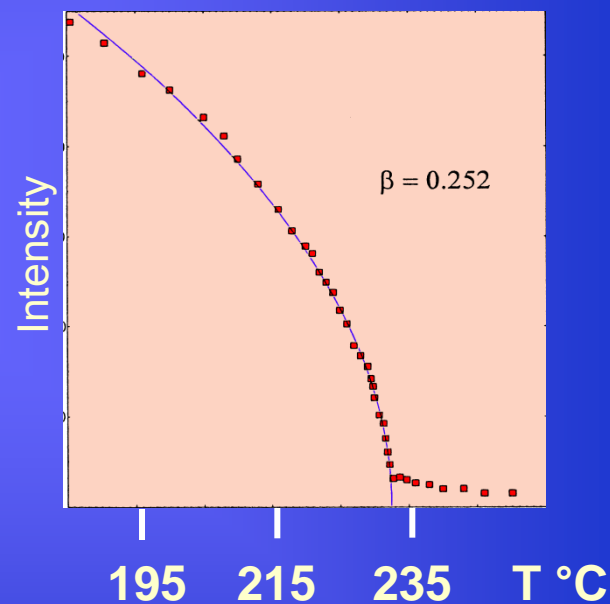
IC-Phase:
An arbitrary superposition
of octagonal rings.



Mechanisms

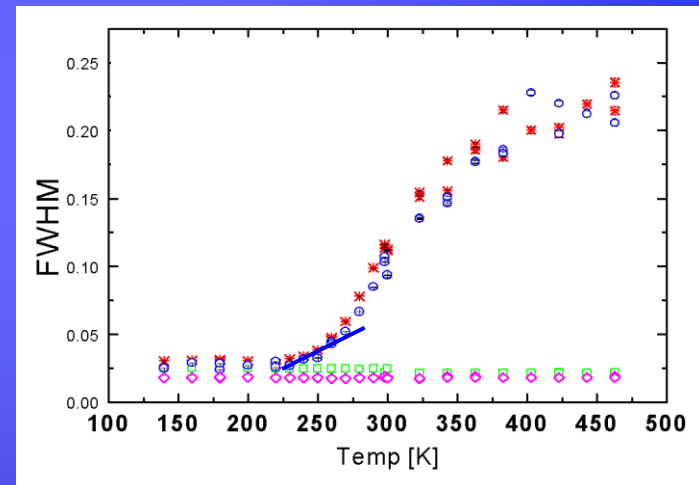
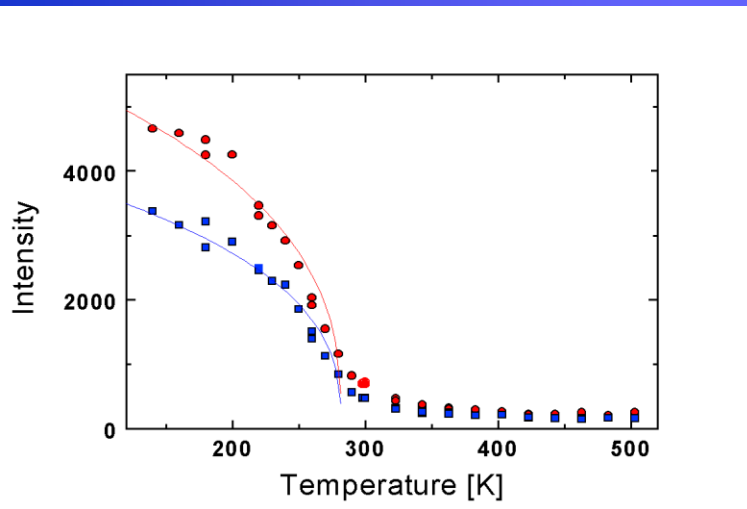
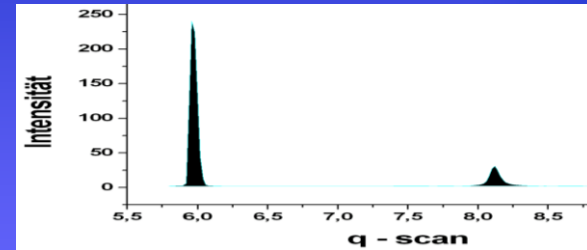
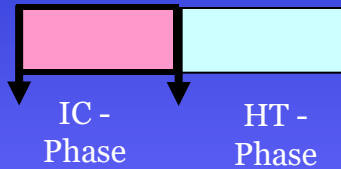
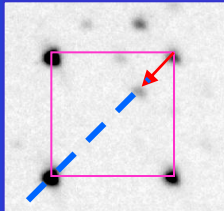


Satellite reflection



Superstructure
reflection

Mechanisms



$$T_c = 277 \text{ K}$$



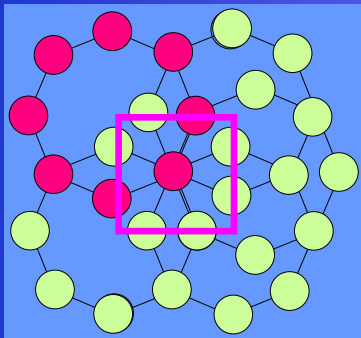
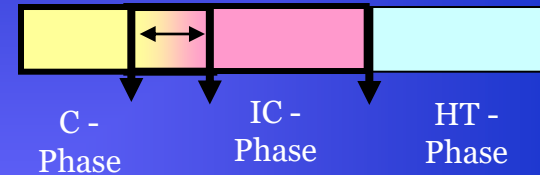
$$T_c = 250 \text{ K}$$

Transformation temperature

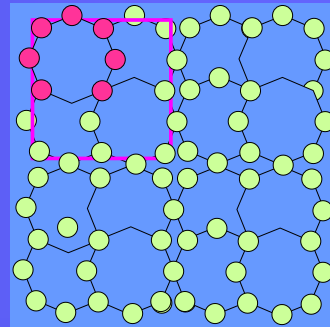
continued



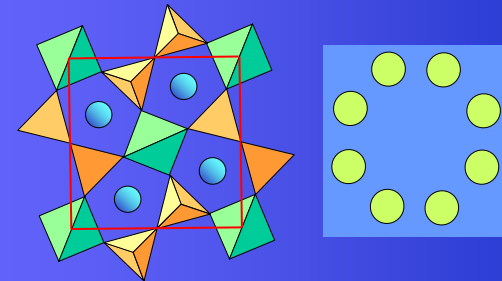
Relationship between the phases



C-Phase:
An ordered
superposition
of octagonal
rings.



IC-Phase:
An arbitrary
superposition
of octagonal
rings.

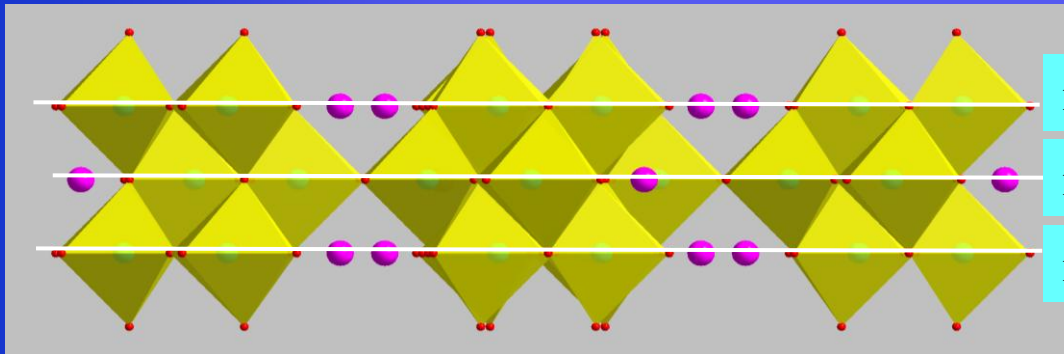


HT-Phase:
Domains of octagonal
rings with short range
order.

continued



structure : β -Phase



m

m

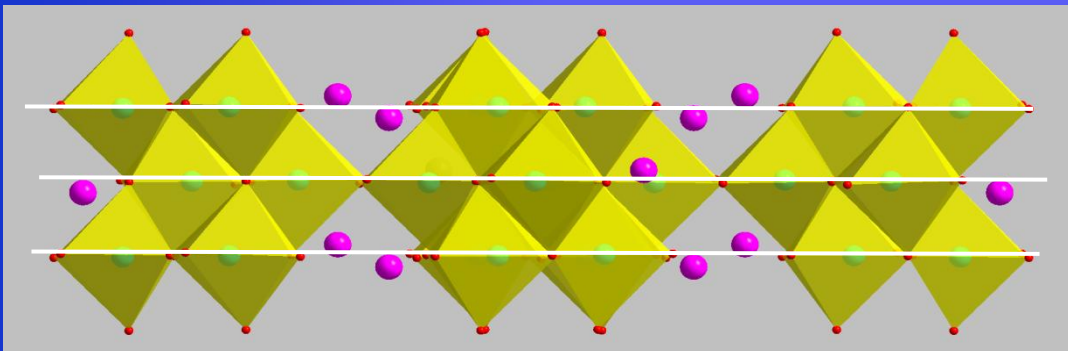
m

Interpretation :

Above T_c the lone pair of Bi-atoms begins to order in domains with short range order.

structure : α -Phase

Bi - Atome



Long range order is attained at the transition point

Final question

What characterizes a continuous phase transition ?

Answer:

The continuous increase of the range of order until long range order is achieved
(not a continuous shift of atomic positions).